

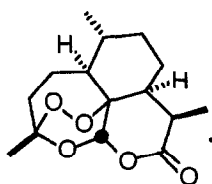
**SYNTHESIS AND ANTIMALARIAL EVALUATION OF
2,3,5-TRIOXABICYCLO[2.2.3]NONANE,
A MODEL FOR THE PUTATIVE PHARMACOPHORE OF ARTEMISININ**

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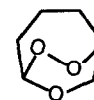
Abstract: The title compound was prepared from the known 6-tetrahydroxepanol by formation of the 6-mesylate followed by treatment with acidic methanol to yield the methyl acetal. Reaction of 2-methoxy-6-mesyloxytetrahydroxepane with anhydrous hydrazine followed by treatment with 30% hydrogen peroxide and sodium peroxide provided the diastereomeric mixture of 6-hydroperoxides. The hydroperoxides cyclized upon treatment with acid to give trioxabicyclo[2.2.3]bicyclononane. The bridged bicyclic trioxane exhibited only marginal antimalarial activity.

The antimalarial drug artemisinin (**1**) has initiated interest in the chemistry and biological activity of related peroxy compounds.¹⁻⁴ The unique 1,2,4-trioxane system embedded in the framework of **1** has been proposed as the structural feature necessary for antimalarial activity.⁵ In our SAR studies of artemisinin, we are pursuing



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an approach that involves the synthesis of suitably substituted hydroperoxides and subsequent cyclization to trioxanes. Through synthesis and antimalarial testing, we intend to clarify whether the bridged trioxane is not only necessary but sufficient to impart antimalarial activity. Our earlier studies have shown that 2,3,5-trioxabicyclo[2.2.2]octanes are only marginally active as antimalarial agents.⁶ We now report the synthesis of the analogous 2,3,5-trioxabicyclo[2.2.3]-

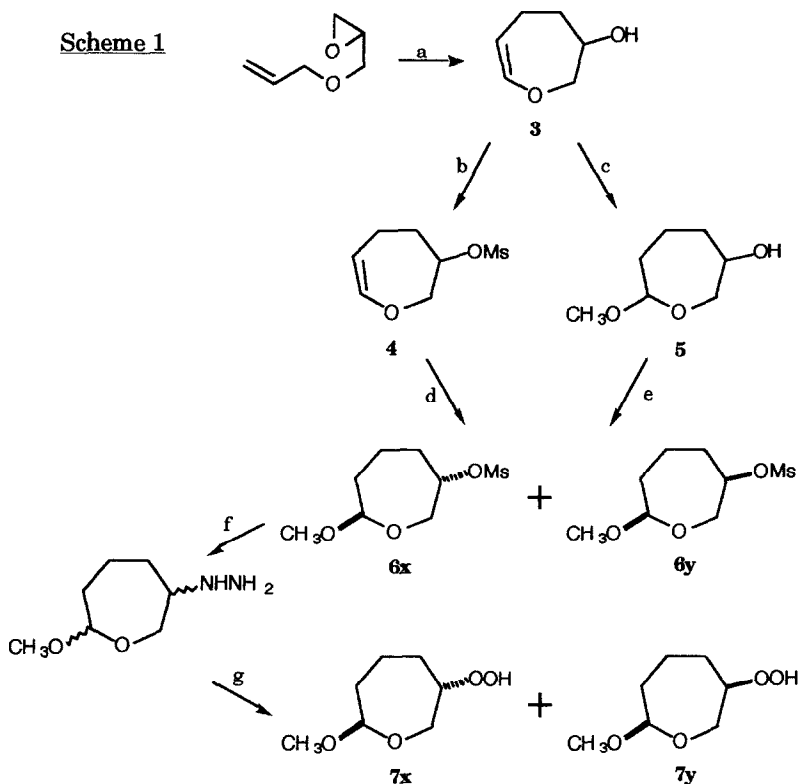


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nonane (**2**) and its activity against malaria parasites.

The synthesis of **2** begins with the known 6-tetrahydroxepanol (**3**), prepared by cyclization of allyl glycidyl ether (Scheme 1).⁷ Two routes were then explored for the preparation of the mesyl acetal (**6**). Compound **4** was formed from **3** using standard conditions and was then treated with methanol and acid to give acetal **6** in 67% yield. Alternatively, compound **3** could be treated with acid and methanol to give **5** (52%) followed by reaction with methanesulfonyl chloride to give **6** (20%).

With compound **6** in hand, the preparation of the corresponding hydroperoxides proceeded using our published protocol.⁸ Mesylate **6** was reacted with anhydrous hydrazine at reflux for several days. After an extractive work-up, the residue was then treated with hydrogen peroxide and sodium peroxide in 2-propanol to provide hydroperoxide **7** which was purified by chromatography in 54% overall yield from **6**.



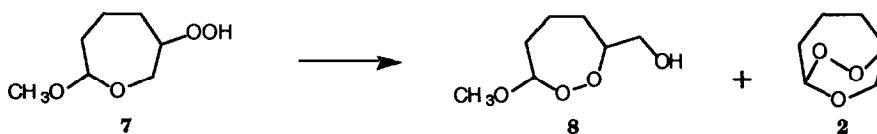
Reaction Conditions: (a) see ref 7; (b) MsCl, Et₃N, 76%; (c) MeOH, Ph₃P•HBr, 52%; (d) MeOH, TsOH, 67%, (e) MsCl, Et₃N, 20%; (f) anhydr. NH₂NH₂, reflux, 4 d, 100% (mass balance); (g) H₂O₂, Na₂O₂, iPrOH, 3d, 54%.

Compounds **6** and **7** exist as mixtures of *cis* and *trans* isomers which can be separated by chromatography and characterized. Although ¹H NMR resonances were assigned by decoupling experiments on compounds **6x** and **6y**, it was not possible to identify the isomers using spectroscopic methods due to the conformational flexibility of the oxepane system. Our studies on the stereochemical issues in these systems will be published elsewhere.

The mixture of *cis* and *trans* hydroperoxides **7** was treated with acid to effect cyclization to the peroxy acetal compound. Some selectivity in the course of the cyclization was observed depending on which acid was used. A number of acids (formic acid, camphor sulfonic acid, Amberlyst 120) that were examined had no effect on **7**. Methanesulfonic acid produced only dioxane **8** in a 40% yield. When Amberlyst-15 ion-exchange resin was used in the presence of molecular sieves, however, mixtures of **8** and the trioxane **2** were isolated in 25% and 21% yields, respectively. Trioxane **2** was isolated in quantitative yield upon reaction of **7** with perchloric acid in CH₂Cl₂.

Both isomers of the hydroperoxide **7** as well as trioxane **2** were screened against *P. falciparum* clones.⁹ The results are shown in the Table along with the data for other active antimalarial agents. As had been seen for other hydroperoxides, the activity in these tests is marginal at best. The trioxane (**2**) shows better activity but not nearly enough to establish the 2,3,5-trioxabicyclo[2.2.3]nonane system as critical for antimalarial activity.

Scheme 2



We are studying the preparation of water soluble analogues of **2** in order to determine whether physico-chemical properties of the trioxanes plays an important role in antimalarial activity.

Experimental

6-Methanesulfonyloxytetrahydrooxepane (**4**)

6-Tetrahydrooxepanol (**3**) (1.2 g, 11 mmol), Et₃N (2.7 mL, 20 mmol), and methanesulfonyl chloride (1.4 mL, 18 mmol) were combined in distilled CH₂Cl₂ (50 mL) and were allowed to stir for 30 min in a dry ice/acetone bath. The mixture was diluted with CH₂Cl₂ (50 mL) and then washed with 2% AcOH (3 x 25 mL). The organic layer was dried (Na₂SO₄) and evaporated under vacuum to give 2.6 g of a colorless oil. The product was purified by chromatography on silica gel using 20% ethyl acetate/hexanes as eluant to provide 1.2 g (76%) of the mesylate. ¹H NMR: 2.20 (br m, 4H, CH₂), 3.06 (s, 3H, OMs), 4.18 (q, 2H, OCH₂), 4.81 (m, 1H, CHOMs), 5.01 (m, 1H, OCH=CH), 6.32 (d, 1H, OCH=CH). ¹³C NMR: 20.80, 31.84, 38.61, 73.60, 80.37, 108.96, 148.83.

2-Methoxy-6-methanesulfonyloxyoxepane (**6**)

6-Methanesulfonyloxytetrahydrooxepane (0.4 g, 2.08 mmol), triphenylphosphine hydrobromide (0.035 g, 0.1 mmol), and distilled dry CH₂Cl₂ (15 mL) were combined under nitrogen. Freshly distilled dry MeOH (0.25 mL, 6.24 mmol) was added via syringe and the reaction stirred for 5 h at RT. After dilution with CH₂Cl₂ (50 mL), the mixture was washed with brine (2 x 25 mL) and the organic layer was separated, dried (Na₂SO₄), and evaporated under vacuum to give a crude oil which was purified by silica gel chromatography to provide 0.3 g (67%) of the mixture of diastereomers.

Band 1: ¹H NMR: 1.44 (m, 1H), 1.63 (m, 3H), 2.06 (m, 1H), 2.30 (m, 1H), 3.02 (s, 3H, OMs), 3.35 (s, 3H, OMe), 3.69 (dt, 1H, OCH₂), 3.89 (dd, 1H, OCH₂), 4.54 (dd, 1H, CHOMe), 4.66 (m, 1H, CHOMs). ¹³C NMR: 17.94, 33.92, 35.43, 38.45, 55.08, 62.94, 79.12, 102.77.

Band 2: ¹H NMR: 1.58 (m, 2H), 1.73 (m, 2H), 2.19 (m, 2H), 3.08 (s, 3H, OMs), 3.36 (s, 3H, OMe), 3.80 (dt, 1H, OCH₂), 4.01 (d, 1H, OCH₂), 4.66 (dd, 1H, CHOMe), 4.90 (d, 1H, CHOMs). ¹³C NMR: 17.43, 34.79, 35.63, 38.87, 55.11, 62.20, 79.66, 102.77.

2-Methoxy-6-hydrazinooxepane

2-Methoxy-6-methanesulfonyloxyoxepane (0.25 g, 1.1 mmol) was heated at reflux in anhydr. hydrazine (5 mL, 96%) for 4 d. The reaction mixture was washed with ether (3 x 20 mL) and the organic layers were washed with 50% aq KOH (5 mL). The ether layer was dried (Na₂SO₄) and evaporated under vacuum to give the crude hydrazino

Table

Compound	Indochina W-2 ^a ng/mL			Sierra Leone D-8 ^a ng/mL		
	IC ₅₀	IC ₉₀	IC ₁₀	IC ₅₀	IC ₉₀	IC ₁₀
7 , isomer A	6524.88	12775.42	3332.49	7490.26	22824.73	2458.03
7 , isomer B	15456.81	20933.22	11413.10	10053.20	38084.48	2653.75
2	--- ^b	--- ^b	--- ^b	2449.32	4466.78	1343.06
artemisinin	0.72	1.05	0.49	0.83	2.13	0.32
chloroquine	70.02	93.12	52.65	2.69	3.61	2.01
mefloquine	2.09	3.73	1.17	6.86	38.47	1.22

^a D-8 is a chloroquine-sensitive strain of *Plasmodium falciparum*; W-2 is chloroquine resistant.
^b Not tested

derivative (0.17 g, quant.) which was used in the next reaction without purification.

2-Methoxy-6-hydroperoxyoxepane (7)

2-Methoxy-6-hydrazinooxepane (70 mg, 4 mmol) was dissolved in 2-PrOH (15 mL) and stirred with 30% aq H₂O₂ (10 mL). Na₂O₂ (50 mg, 0.6 mmol) was added and the reaction allowed to stir for 3 d. The reaction mixture was diluted with CH₂Cl₂ (100 mL) and washed with brine (2 x 50 mL). The organic layer was dried (Na₂SO₄) and evaporated under vacuum to give a yellow oil. The crude hydroperoxides were purified on silica gel to give both diastereomeric hydroperoxides (54 %).

Band 1: ¹H NMR: 4.53 (q, 1H, CH-O-C), 4.06 (m, 1H, CH-OOH), 3.85 (d, 2H, CH₂-O-C), 3.4 (s, 3H, OCH₃), 2.10-2.0 (m, 2H, CH₂), 1.75-1.65 (m, 2H, CH₂), 1.4-1.25 (m, 2H, CH₂). ¹³C NMR: 103.57, 83.67, 62.17, 55.04, 34.12, 31.94, 18.26. MS: 145 (M-OH), 131.1 (M-MeO).

Band 2: ¹H NMR: 8.44 (br s, 1H, OOH), 4.62 (q, 1H, CH-O-C), 4.13 (m, 1H, CH-OOH), 3.93 (d, 2H, CH₂-O-C), 3.36 (s, 3H, OCH₃), 2.09-2.03 (m, 2H, CH₂), 1.68-1.26 (m, 4H, CH₂). ¹³C NMR: 102.87, 81.88, 59.79, 55.01, 34.79, 32.29, 17.85.

7-Hydroxymethyl-3-methoxydioxepane (8)

Compound 7 (10 mg, 0.06 mmol) was dissolved in freshly distilled CH₂Cl₂ (20 mL). Molecular sieves (4Å, 100 mg) and methanesulfonic acid (1 drop) were added. After stirring at room temperature, the reaction mixture was diluted with CH₂Cl₂ and washed with satd NaHCO₃. The organic layer was separated, dried (Na₂SO₄) and evaporated to give an oil which was purified by chromatography on silica gel to provide the dioxepane (4 mg, 40%) as a mixture of diastereomers (~6:1). ¹H NMR: 1.39-2.14 (m, 6H), 3.46 (s, 3H, OMe), 3.50 (m, 2H, CH₂OH), 4.32 (m, 1H, CHOO), 4.74 (q, 1H, OCHO). ¹³C NMR: 19.71, 20.93, 30.68, 32.16, 33.54, 34.70, 53.41, 55.62, 62.97, 63.55, 84.19, 87.49, 108.71.

2,3,5-Trioxobicyclo[2.2.3]nonane (2)

A mixture of the diastereomers of 2-methoxy-6-hydroperoxyoxepanes (50 mg, 0.3 mmol) dissolved in dry CH₂Cl₂ (25 mL) with molecular sieves (500 mg) was stirred with 2 drops of conc. HClO₄. After 30 min, when by TLC there was no starting material, the reaction mixture was filtered and diluted with CH₂Cl₂ (50 mL). It was then washed with satd. NaHCO₃ (2 x 20 mL) and dried with Na₂SO₄. Evaporation of the CH₂Cl₂ under vacuum gave the product as a white solid, mp 160-161 °C (40 mg, 100%). ¹H NMR: 5.13 (br s, 1H, acetal), 4.35 (t, 1H, CH-OOC), 3.95 (d, 1H, endo to endoperoxide), 3.52 (q, 1H, exo to endoperoxide), 1.87 (br s, 2H, CH₂), 1.58 (br s, 2H, CH₂), 1.35 (br s, 2H, CH₂). ¹³C NMR: 103.80, 79.21, 69.13, 31.71, 28.95, 19.23. MS: 131.1 (M+1, 18%), 85.1 (M-46, 100%)

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References

- (1) Coordinating Group for Research on the Structure of Qing Hau Sau *K'o Hsueh T'ung Pao* **1977**, *22*, 142.
- (2) China Cooperative Research Group on Qinghaosu and its Derivatives as Antimalarials *J Trad Chin Med* **1982**, *2*, 3.
- (3) Klayman, D. L. *Science* **1985**, *228*, 1049.
- (4) Luo, X. D.; Shen, C. C. *Med Res Rev* **1987**, *7*, 29.
- (5) Kepler, J. A.; A., P.; Lee, Y. W.; Musallam, H. A.; Carroll, F. I. *J Med Chem* **1987**, *30*, 1505-1509.
- (6) Casteel, D. A.; Jung, K.-E.; Gerena, L.; Milhous, W. *Bioorg Med Chem Lett* **1992**, *2*, 623-626.
- (7) Bird, C. W.; Hormozi, N. *Tetrahedron Lett* **1990**, *31*, 3501-3504.
- (8) Casteel, D. A.; Jung, K.-E. *J Chem Soc., Perkin Trans. 1* **1991**, 2597-2598
- (9) See ref 6 for details of the screening experiments.